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Chapter 20

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Reduction

Reactions

Answers

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~~Oxidation and~~

~~Reduction~~

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1 of 2 Chapter
20 — Oxidation
Reactions
and Reduction~~

~~Answers: Part
1 of 5 Chapter
20 —~~

~~Electrochemistry
Oxidation and
Reduction~~

**Oxidation and
Reduction
Reactions -
Basic**

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Introduction

~~Chapter 20 —
Oxidation and
Reduction~~

~~Reactions: Part
2 of 2 Chapter
20 — Oxidation
and Reduction~~

~~Reactions: Part
4 of 5~~ **Chapter**

20

Electrochemistry

*Ch 20 balancing
oxidation*

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Oxidation

Practice

*Questions for
Chapter 20*

Chapter 20 -

*Oxidation and
Reduction*

Reactions: Part

2 of 5 **Mr Z AP**

Chemistry

Chapter 20

lesson 1: Redox

Equations and

Oxidation

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Numbers

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Oxidation and
Reduction

Reactions: Part

5 of 5 ~~Oxidation -
Reduction~~

~~Reactions~~ GCSE

Chemistry -

Oxidation and

Reduction -

Redox Reactions

#32 (Higher

Tier) 08. Oxidat

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~~Oxidation-Reduction~~

~~Reactions~~

~~Electrochemistry~~

~~Review – Cell~~

~~Potential \u0026~~

~~Notation, Redox~~

~~Half Reactions,~~

~~Nernst Equation~~

~~What is the~~

~~Difference~~

~~Between~~

~~Oxidation \u0026~~

~~Reduction +~~

~~Types of~~

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~~Chemical
Reactions +
Chemistry
Electron~~

~~Exchange In Oxid
ation-Reduction
Reactions /
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Chemistry #36**

Special Topics 1

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Review of Gen-
Chem: Part 6 of
15 Introduction
to

Electrochemistry
Oxidation and
reduction |

Redox reactions
and

electrochemistry

| Chemistry |

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Chemistry
Chapter 20:
Lesson 1: The
Meaning of
Oxidation and
Reduction
Chapter 20
Oxidation and
Reduction
Reactions: Part
3 of 5 chapter
20 lecture 20 LO
1 and 2~~

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~~REDOX REACTIONS~~

~~(CH_20) Oxidation
vs. Reduction,
What are~~

~~Oxidation and~~

~~Reduction~~

~~Reactions in~~

~~Everyday Life?~~

CHEM 2320

Chapter 20

2-27-19 ~~Chapter~~

~~20 — Electrochem~~

~~istry: Part 1 of~~

~~13~~ Chapter 20 -

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Electrochemistry

• Part 2 of 13

Chapter 20

Oxidation

Reduction

Reactions

Chapter 20

Outline

Oxidation

Reduction

Reactions

Section 20.1 -

Oxidation vs.

Reduction

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Oxidation reactions are also known as reduction reactions. is the loss of electrons or the gain of electrons. is the loss of electrons or the gain of electrons. The way to remember the difference in oxidation and reduction is.

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Outline.docx -

Chapter 20

Outline

Oxidation ...

balancing a

redox equation

by comparing the

increase and

decrease in

oxidation

numbers: half

...

Quia - Chapter

Page 18/48

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20 "Oxidation-
Reduction
Reactions"

oxidation

reduction

reactions

chapter 20

Flashcards. A

molecule is

oxidized when

it... A molecule

is reduced when

it.... a

reaction that

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involves the
transfer of
electrons
between rea... 1)

donates
electrons (fully
or in a covalent
bond)... 2)

gains ox... 1)
gains electrons
(fully or in a
covalent
bond) ...

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*oxidation
reduction
reactions
chapter 20*

Flashcards and

Pearson Lesson

Chapter 20 Notes

Oxidation-

Reduction

Reactions 20.1

The Meaning of

Oxidation and

Reduction What

are Oxidation

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and Reduction? o

Oxygen and Redox

KEY = The

substance

gaining O is

oxidized, while

the substance

losing O is

reduced Oxidatio

n-Reduction

Reactions =

Reactions with a

substance being

oxidized and

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another

Reduction

Chapter 20 Notes

Oxidation

Reduction

Reactions

Chapter 20 -

Oxidation and

Reduction

Reactions: Part

1 of 2 Mike

Christiansen.

... Oxidation

Reduction

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Oxidation -

Redox -

Duration: 3:33.

Brightstorm

146,723 views.

3:33.

Chapter 20 -

*Oxidation and
Reduction*

*Reactions: Part
1 of 2*

? Reduction

reactions are

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the opposite of
oxidation ?
Originally, this
was believed to
signify simply
the loss of
oxygen from a
compound ? That
is a good rule
of thumb, but is
not always the
case ? A common
example is the
reduction of

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iron ore?

Oxygen is removed, iron ore and carbon

dioxide are

formed? This occurs when iron ore and carbon

are heated together

Chapter 20:

Oxidation

-Reduction

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Oxidation

trons also is an oxidation-reduction reaction.

636 Chapter 20

Redox Reactions

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Figure 20-1 The reaction of magnesium and oxygen involves a transfer of electrons from magnesium to oxygen.

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Therefore, this reaction is an oxidation-reduction reaction.

Using the classifications given in Chapter 10, this redox reaction also is classified as a combustion reaction. X

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Redox Reactions

biochemical oxidation-reduction reactions the

transfer of

hydrogen atoms is necessary for the production

of energy in the

cells. methyl

alcohol (CH_3OH)

a poisonous

substance, is

metabolized in

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the body by the
following
reactions: CH_3OH
 $\rightarrow \text{H}_2\text{CO} + 2\text{H}$

Answers

Chapter 20:
Redox reactions
Flashcards /

Quizlet

Chapter 20

Worksheet: Redox

ANSWERS I.

Determine what
is oxidized and

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what is reduced
in each
reaction.

Identify the
oxidizing agent
and the reducing
agent, also. 1.

$2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$ Sr
0 to Sr^{2+} ; oxidized/reducing
agent O_0 to O^{2-} ;
reduced/ox. ag.

2. $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$
Li 0 to Li^{+} ;

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oxidized/red.

ag. S^0 to S^{2-} ;

reduced/ox. ag.

3.

Answers

Chapter 20

Worksheet Redox

- Beverly Hills

High School

Chapter 20 Redox

Reactions -

Notes. Redox

reaction. - a

reaction in

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which electrons are transferred from one atom to another.

Oxidation. -

loss of electrons from atoms of a

substance. Ex -
 $\text{Na} \rightarrow \text{Na}^+ + \text{e}^-$.

Sodium is oxidized.

Reduction. -

gain of

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oxidation by
atoms of a
substance.

Reactions

Chapter 20 Redox

Reactions -

Notes

Chapter 20 20-1

Chapter 20

Electrochemistry

•

Electrochemistry

deals with the

relationships

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Oxidation

electricity and
chemical

reactions. • Oxi

dation-reduction
(redox)

reactions were

introduced in

Chapter 4 • Can

be simple

displacement

reactions: $\text{Zn}(s)$

+ $\text{Cu}^{2+}(aq)$?

$\text{Zn}^{2+}(aq)$ + $\text{Cu}(s)$

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$\text{Cu(s)} + 2\text{Ag}^+(\text{aq})$
 $\rightarrow \text{Cu}^{2+}(\text{aq}) +$
 2Ag(s) 20.1 Oxidation-Reduction
Reactions

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Chapter 20:
Oxidation States
and Redox
Reactions Electr
ochemistry: the
study of the
interchange of

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Chemical and
electrical
energy Review
Oxidation

reduction

reactions

involve a

transfer of

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the end of your
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chapter 20.1-2
oxidation states

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*and redox
reactions.docx*

•••
Oxidation-

Reduction

Reactions

Chemistry

Chapter 20 -

Redox Reactions

That Form Ions

(Ionic

Compounds) In

metal / nonmetal

reactions,

Page 38/48

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Oxidation reactions are
transferred from
the metal atom
to the nonmetal.

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presentation |
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*PPT - Chapter 20
OxidationReducti
on Reactions
Redox ...*

CHAPTER 20 " Oxi
dation-Reduction

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Oxidation” LEO

SAYS GER. 2.

Section 20.1 The
Meaning of

Oxidation and

Reduction

(called “redox”)

OBJECTIV
ES Define
oxidation and
reduction in
terms of the
loss or gain of

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oxygen, and the
loss or gain of
electrons.

.

3. Answers

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Chemistry - Chp

20 - Oxidation

Reduction

Reactions ...

Chapter 14 -

Gases; Chapter

15 - Solutions;

Chapter 16 -

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Oxidation
Energy and
Chemical Change;
Reduction
Chapter 17 -
Reactions
Reaction Rates;
Answers
Chapter 18 -
Equilibrium;
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Chapter 19 -
Check
Acids and Bases;
Chapter 20 -
Redox Reactions;
Chapter 21 - Ele
ctrochemistry;
Chapter 22 -
Hydrocarbons;

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Chapter 23 -
Substituted
Hydrocarbons and
Their Reactions;
Chapter 24 - The
Chemistry of ...

Chapter 20 -
Redox Reactions
- Ms. K Kelly -
John F ...

CHAPTER 20 Oxida
tion-Reduction
Reactions LEO

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Oxidation Using
half-reactions
continued Step
4: add enough
electrons to one
side of each
half-reaction to
balance ... - A
free PowerPoint
PPT presentation
(displayed as a
Flash slide
show) on
PowerShow.com -

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Oxidation Reduction

PPT - Chapter 20

Oxidation

Reduction

Reactions

PowerPoint ...

Chapter 20 - Oxi
dation-Reduction
Reactions - 20.1

The Meaning of
Oxidation and
Reduction -

Sample Problem

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20.1 – Page 695:

1 Answer Na is oxidized and is the reducing agent.

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20 - Oxidation-
Reduction ...

Oxidation-
reduction
reactions.

Oxidation and

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reduction. This is the currently selected item.

Oxidation state trends in periodic table.

Practice determining oxidation

states. Unusual oxygen oxidation states.

Balancing redox equations.

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Oxidizing and
reducing agents.
Disproportionati
on.

Answers

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