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Chemistry Chapter 6 Study Guide. Lily Taylor. 19 October 2020 .
question. period. answer. Horizontal row in the periodic table.
question. group. answer. vertical column in the periodic table.
question. periodic law. answer. repetition of properties occurs when

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elements are arranged in order of increasing atomic number.

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properties periodic law atomic number Henry Moseley Dmitri Mendeleev
The table below was developed by John Newlands and is based on a
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6.1 Development of the Modern Periodic Table. 1.octaves. 2.eight.
3.nine. 4.accepted. 5. Dmitri Mendeleev. 6.atomic mass.

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Chemistry Chapter 6 Study Guide. arranged elements according to atomic
mass and used the arrangement to predict the properties of missing
elements. elements that are characterize by the filling of p orbitals
are classified as _____. which subatomic particle plays the greatest
part in determining the properties of an element.

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with water.

Name

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should be able to... Define chemistry. Describe the difference between mass and weight. Describe the difference between chemical and physical change. Describe the difference between accuracy and precision. Describe the five states of matter, and give an example of each

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